



MECHANISM OF VITAMIN B₁₂ MALABSORPTION IN BLIND LOOP SYNDROME: BACTERIAL UPTAKE OF VITAMIN B₁₂ AND STABILITY OF BINDING OF VITAMIN B₁₂ TO GASTRIC INTRINSIC FACTOR*¹

BY

Takashi NAKAJIMA, Tsukasa ABE and Hiroyoshi OKAMURA*²

ABSTRACT

Both free vitamin B₁₂ (B₁₂) and B₁₂ bound to the intrinsic factor (IF) were taken up by the microorganisms. The amount of B₁₂ taken up in the presence of IF was fairly smaller than in the absence of IF.

The stability of the B₁₂-IF complex was influenced by the temperature as well as by pH. The bacterial uptake of B₁₂ appeared to be favourable under the condition where the vitamin dissociated easily from IF. Moreover, when the B₁₂-IF complex was incubated with the microorganisms, the free IF activity in the supernatant increased according to the bacterial uptake of B₁₂.

It seemed that the microorganisms took up B₁₂ after the liberation of B₁₂ from IF. The amount of B₁₂ available for bacterial uptake was regulated by the rate of dissociation of the B₁₂-IF complex as well as by the reciprocal changes in the size of the bacterial population and IF available.

It is suggested that B₁₂ malabsorption in the blind loop syndrome could be partly explained by the bacterial uptake of B₁₂ in the presence of abundant microorganisms in the loop and by the predominant dissociation of the vitamin from IF.

INTRODUCTION

The mechanism of B₁₂ malabsorption in the patients with the blind loop syndrome has been obscure. Microorganisms in the loop are considered to play an important role in this malabsorption. (i) Malabsorption of B₁₂ in patients with blind loop was improved immediately after the administration of antibiotics and reappeared by the cessation of the antibiotics¹⁻¹¹. (ii) The microorganisms in the loop took up the vitamin in vitro^{5,12-20}. (iii) Malabsorption of B₁₂ was also observed in the experimental animals with diverticula²¹⁻²³. (iv) It was reported that the mor-

*¹ A preliminary report of this work was presented at the autumn meeting of the Japanese Society of Gastroenterology, Oct. 17, 1970.

*² 中島 隆, 阿部 帥, 岡村裕喜: The First Department of Internal Medicine (Chief: Prof. M. KOMIYA), Tokyo Medical and Dental University (Tokyo Ika Shika Daigaku). Received for publication, August 1, 1972.

phological changes of the intestinal mucosa with a blind loop in the patients^{5,8,9,24}) and in the animals²¹⁻²³) were not so severe as those by which B₁₂ malabsorption was brought about. Furthermore, B₁₂ uptake by the isolated, perfused segment of the intestine was the same in the rat with diverticula as was observed in the control animals²³).

It has been reported that the bacterial uptake of B₁₂ initially bound to the gastric intrinsic factor was negligible or remarkably reduced¹²⁻²⁰). Dellipiani et al.¹⁷) reported that human gastric juice had a marked inhibitory effect on the B₁₂ uptake by *E. coli*, but it was an incomplete inhibition. Although significant amounts of bound B₁₂ are taken up by the intestinal microorganisms in vitro, the regulatory mechanism of B₁₂ uptake has not been elucidated, especially when the activity of the intrinsic factor remained unchanged. It is essential to understand the role of the microorganisms in B₁₂ uptake for the investigation of B₁₂ malabsorption. Present investigation was designed to solve this mechanism.

MATERIALS AND METHODS

Microorganisms Typical B₁₂ malabsorption was observed in two patients with a blind loop which was made by anastomosing the middle portion of the small intestine to the transverse colon. Microorganisms including *E. coli*, *Klebsiella* and *Citrobacter* were obtained from the contents of the blind loop, when the operation to liberate the anastomosis was performed for radical treatment of the malabsorption without pre-administration of antibiotics. Some of the standard strains of *E. coli* and several strains of *Klebsiella* and *Citrobacter* from other origins in other patients were also used. These microorganisms were preserved in heart infusion agar media. For examinations, the microorganisms were inoculated twice in heart infusion broth, washed twice with distilled water or saline, and resuspended in these aquae. Usually a certain amount of microorganisms was measured by their turbidity in the final heart infusion broth. The turbidity was determined by multiplying the photometric reading at a wave length of 600 m μ by 1,000. The bacterial suspension with a turbidity from 300 to 400 contained approximately 10⁷ to 10⁸ microorganisms per ml.

When bacterial suspensions were incubated at temperatures between 0°C and 45°C for 4 hours, the B₁₂-binding capacity of these supernatants was below 0.002 u per tube. The fresh broth contained approximately 2.5 m μ g/ml of B₁₂, while B₁₂ was detected sparsely in the supernatant after incubation with the microorganisms.

Cyanocobalamin (B₁₂) ⁵⁷Co-labelled cyanocobalamin (⁵⁷Co-B₁₂), 10 μ Ci

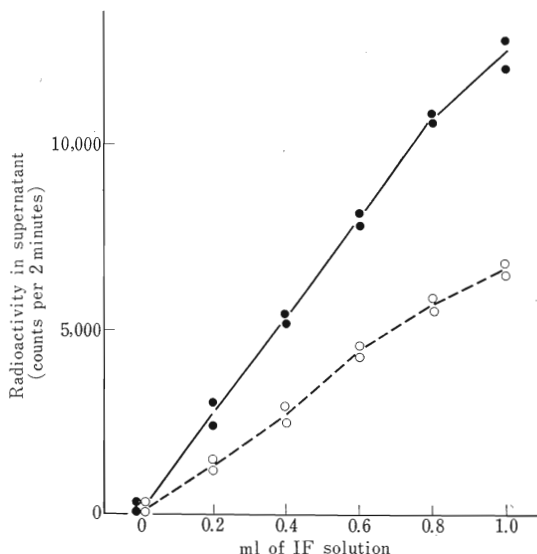


Fig. 1. Separation of bound B₁₂ from free B₁₂. After incubating B₁₂ (0.089 m μ g) and IF (0.087 u/ml) with bovine albumin and adding coated charcoal, the radioactivity in the supernatant was measured by 2-minute counting.

Closed circles: original solution of ⁵⁷Co-B₁₂

Open circles: mixture of equal amount of ⁵⁷Co-B₁₂ and cold B₁₂

per 82 to 89 m μ g per ml, was obtained from the Radiochemical Centre, Amersham, England and diluted with nonradioactive B₁₂ to the desired concentration. The radioactivity was counted in duplicated samples for 2 minutes by the well-type scintillation counter (Spectro Scaler RDM-1, Tokyo Shibaura Electric Co., Japan).

Bovine albumin-coated charcoal (coated charcoal) The activity of the intrinsic factor (IF) and antibody to IF (IFA) were measured by the "coated charcoal" method of Gottlieb et al.²⁵⁾ In the present investigation, 1 unit (u) of IF represented the B₁₂-binding capacity of IF which bound 1 m μ g of B₁₂, and 1 u of IFA (blocking antibody) was equivalent to 1 u of IF.

Using the coated charcoal, the bound B₁₂ was separated from the free B₁₂, and the solutions of "B₁₂-saturated IF" were prepared. The separation of bound B₁₂ from free B₁₂ was successful, even when 0.089 m μ g of B₁₂ and small doses of IF were incubated with approximately 0.01 ml of 1% bovine albumin (Fig. 1). The recovery of IF by the above procedure was from 87 to 91% of the expected values. B₁₂-binding capacity of the sera with IFA, of the supernatants of bacterial suspensions and of the bovine albumin

was also determined with 0.089 m μ g of B₁₂ and 1 ml of coated charcoal. B₁₂-binding capacity of 1% bovine albumin was below 0.01 u/ml.

Gastric intrinsic factor (IF) Gastric juice was obtained from hypersecretors after an administration of 50 mg of betazole hydrochloride. The obtained juice contained IF which was responsible for more than 98% of the total unsaturated B₁₂-binding capacity. It was used as the IF solution which was preserved at -20°C until required.

Antibody to intrinsic factor (IFA) IFA was obtained from the sera of two patients with pernicious anemia. Titters of the "blocking antibody" were 575 and 786 u/ml. These sera also had a "binding antibody" and a B₁₂-binding capacity of approximately 0.1 u/ml.

B₁₂-saturated IF After mixing 1 volume of IF and approximately 3 volumes of B₁₂, coated charcoal was added and followed by centrifugation. The supernatant was employed as the "B₁₂-saturated IF".

Bacterial uptake of B₁₂ Bacterial suspensions were incubated with either free or bound B₁₂, and the total volumes of the mixtures were from 3 to 4 ml. After the incubated mixtures were centrifuged, the sediments were washed twice with distilled water or saline. The amount of the B₁₂ taken up by the microorganisms ("bacterial B₁₂") was measured by counting the radioactivity of the sediment or the supernatant. The "B₁₂-binding capacity of the microorganisms" usually indicated the "bacterial B₁₂" when the microorganisms (turbidity 1,000, 1 ml) were incubated at 37°C for 1 hour with 100 m μ g of free B₁₂.

Nature of B₁₂ taken up by microorganisms In order to examine the possibility that microorganisms directly took up bound B₁₂, IF was incubated at 37°C for 4 hours with various amounts of microorganisms preincubated in the presence or absence of B₁₂. The decrease in the IF activity in the supernatant suggests that the microorganisms take up (or bind with) bound B₁₂. Cold B₁₂-saturated IF (0.125 u) was incubated with *E. coli* (turbidity 270, 5 ml). After this, the supernatant fraction was incubated with and without IFA (1.5 u) and labelled free B₁₂ (0.125 m μ g) was added, followed by the administration of 1 ml of coated charcoal. The labelled B₁₂ in the final supernatant was measured, and the free IF activity was expressed by the difference between the amount of the labelled B₁₂ with and without IFA. Simultaneously, the "bacterial B₁₂" was measured with labelled B₁₂-saturated IF (0.125 u). The increase in free IF in the supernatant according to the increase in the "bacterial B₁₂" suggests that bound B₁₂ is taken up by the microorganisms after the liberation of B₁₂ from IF.

Stability of B₁₂-IF complex The exchange between free B₁₂ and bound B₁₂, or the dissociation of B₁₂ from IF, was examined by the following

methods with a single isotope: *Experiment A*: Labeled B₁₂-saturated IF (0.05 u) and cold B₁₂ (100 m μ g) were incubated at varying temperatures between 0°C and 45°C for the period up to 5 hours. Each mixture was cooled immediately after the incubation and then 1 ml of coated charcoal was added. The amount of B₁₂ dissociated from the bound B₁₂ was calculated by the radioactivity of the labelled B₁₂ which was adsorbed on the coated charcoal. *Experiment B*: Mixtures employed were cold B₁₂-saturated IF (0.05 u) and labelled free B₁₂ (0.12 m μ g). Procedures of the experiment were the same as in experiment A, except that the radioactivity was counted in the supernatants. Theoretically, 29.4% radioactivity in the supernatant indicated the rate of complete exchange of the labelled B₁₂ between free and bound B₁₂. *Experiment C*: Mixtures of labelled B₁₂-saturated IF (0.05 u) and IFA (1.50 u) were employed. The radioactivity adsorbed on to the charcoal was determined to be equivalent to the amount of the labelled B₁₂ dissociated from IF.

Effects of pH The stability of the B₁₂-IF complex was examined by changing the pH. The incubation medium was adjusted to various pH from 2.4 to 10.3 with phosphate-buffered saline. The mixtures of labelled B₁₂-saturated IF (1.4 u) and cold B₁₂ (100 m μ g), and the mixture of cold B₁₂-saturated IF (1.4 u) and labelled B₁₂ (1.0 m μ g) were incubated with this medium at 37°C for 90 minutes. The rate of dissociation of B₁₂-IF complex was calculated as mentioned above. The bacterial uptake of B₁₂ was also examined in relation to pH. In these examinations, the pH in the whole incubation mixture was not different from the pH in the stock solution of the phosphate-buffered saline as shown by the pH test paper.

Destruction of B₁₂-saturated IF and IF The B₁₂ adsorbed on to the coated charcoal was estimated by the amount of the B₁₂ liberated from the B₁₂-saturated IF or the amount destroyed, when the B₁₂-saturated IF and coated charcoal were mixed. The destruction did not exceed 10% in the experiment herein described. The loss of free IF activity was examined by measuring the B₁₂-binding capacity of IF. The rate of IF destruction was the same or somewhat greater than that of the B₁₂-saturated IF.

The results presented in this report were not calibrated by the amount of destruction of IF or the B₁₂-saturated IF.

RESULTS

Part 1. Uptake of free B₁₂ by microorganisms

(a) *E. coli* and *Klebsiella*, the turbidity of these being adjusted to 1,000, were suspended in 1 ml of distilled water. The "B₁₂-binding capacity

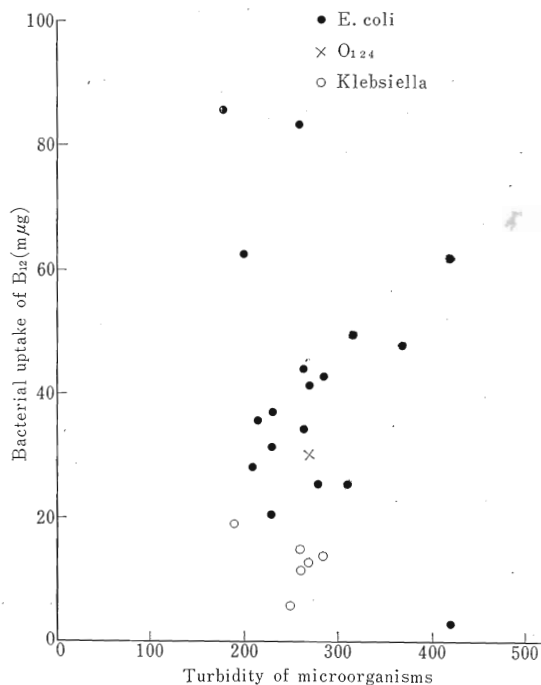


Fig. 2. Bacterial uptake of free B₁₂. *E. coli* and *Klebsiella* (turbidity 1,000, 1 ml) were incubated with 100 mµg of B₁₂ at 37°C for 1 hour and their B₁₂-binding capacity was measured. The abscissa shows the turbidity of the microorganisms in the final broth.

of microorganisms" is shown in Fig. 2. The amount of B₁₂ taken up by *E. coli* was to some extent greater than that by *Klebsiella*. One of the standard strains of *E. coli* (O₁₂₄), which had no cillium, had the B₁₂-binding capacity of 30 mµg. This was rather low among the *E. coli* examined. The vitamin taken up by the microorganisms was not removed by repeated washings, incubating at 37°C for 26 hours or mixing with cold B₁₂ or with large quantities of IF as shown in Fig. 4.

(b) The "bacterial B₁₂" slightly increased along with the increased amount of B₁₂ in the incubation medium, when incubated at 37°C for 1 and 4 hours with 100 to 10,000 mµg of labelled B₁₂ with the same radioactivity. It was not observed that the ⁵⁷Co was prone to be taken up by the microorganisms when incubating with 100 mµg of labelled B₁₂ with different radioactivity. When the microorganisms (turbidity 660, 1 ml) were incubated in 1, 2 and 4 ml of distilled water, the B₁₂-binding capacity of these increased slightly as the volume of the incubation medium increased.

Table 1. Bacterial uptake of free B₁₂. *Klebsiella* (turbidity 330) was incubated with 100 and 0.76 m μ g of B₁₂. Percent uptake of B₁₂ was expressed as the mean of duplicated data.

Incubation temperature	Bacterial suspension (ml)	Incubation time (minutes)						
		(100 m μ g of B ₁₂)				(0.76 m μ g of B ₁₂)		
		10	60	120	240	10	60	120
37°C	1	5.0	7.3	9.5	13.7	72.9	89.0	89.4
	2	7.8	9.6	14.2	18.7	87.7	89.5	91.5
	4	11.7	14.4	18.8	25.4	91.9	93.3	93.9
25°C	1	3.1	5.1	7.0	10.0	61.8	82.7	90.3
	2	5.3	7.7	9.2	12.9	80.2	88.4	90.4
	4	8.7	11.0	14.1	17.8	88.3	89.3	90.2
2°C	1	0.5	0.4	0.5	0.5			
	2	0.6	0.7	0.7	0.7			
	4	1.2	1.2	1.2	1.3			

(c) The B₁₂ uptake by *Klebsiella* was measured by changing the concentration of the B₁₂ and microorganisms or by changing the incubation time and temperature (Table 1). The "bacterial B₁₂" increased as the incubation temperature was elevated. The "bacterial B₁₂" increased gradually as the incubation time was prolonged, not only by incubation with 100 m μ g of B₁₂ but also with a very small dose of B₁₂ (0.076 m μ g). The B₁₂ uptake by *E. coli* was also observed by incubating with 100 m μ g of B₁₂ at 37°C for 1, 5, 16 and 26 hours. They were 14.2, 27.9, 57.3 and 65.1 m μ g, respectively. The "bacterial B₁₂" increased along with the increase in bacterial population. The increase in "bacterial B₁₂" was slightly lower than the value expected from the increase in bacterial population, even when a large dose of B₁₂ (100 m μ g) was used for a short time (10 min.). The same property was found in the "bacterial B₁₂" at time zero, which was suspected on the basis of the "bacterial B₁₂" at 10, 60 and 120 minutes.

(d) The ⁵⁷Co-B₁₂-binding capacity of *E. coli* (turbidity 210, 3 ml) was measured, after pre-incubation with 0, 10, 50 and 100 m μ g of cold B₁₂ at 37°C for 1 hour and washing immediately. It was 26.2, 22.7, 22.5 and 19.5 m μ g, respectively.

(e) The microorganisms were inoculated three times in the minimal medium without B₁₂. After each inoculation, the microorganisms were again inoculated in the minimal medium with and without B₁₂ (5 m μ g/ml). The growth of these was not influenced by the addition of B₁₂.

(f) There was no significant difference in the B₁₂-binding capacity between the microorganisms obtained in the early logarithmic phase of growth and those obtained at the later stage. The B₁₂-binding capacity of

Table 2. Bacterial uptake of free and bound B_{12} . After mixing B_{12} and IF for 10 minutes at room temperature, the mixtures were incubated with microorganisms at 37°C for 30 minutes.

Experiment	B_{12} (m μ g)	IF (u)	Microorganism (ml)	Bacterial uptake of B_{12} (%)	
				Klebsiella	E. coli
1	1	0	2	90.5	85.4
2	1	5	2	1.3	0.9
3	1	5	8	4.3	2.4
4	1	5	40	17.3	13.6
5	1	25	40	8.6	4.9
Turbidity of microorganisms				870	620

the microorganisms was reduced to approximately 70 to 80% of the initial values, when the bacterial suspension in distilled water or saline was kept at 37°C or at room temperature for 3 hours. Those microorganisms had little or no B_{12} -binding capacity after they were soaked in acetone, 8% formalin or 70% ethanol, or when heated at 80°C for 10 minutes.

(g) In the present observation, there were approximately 10% of ^{57}Co which could not be taken up by the microorganisms or by the IF, being adsorbed on to the coated charcoal. No further study was made on this point.

Part 2. Uptake of bound B_{12} by microorganisms

(a) Bound B_{12} was also taken up by the microorganisms, although the "bacterial B_{12} " was diminished remarkably as compared with those without IF (Table 2). The "bacterial B_{12} " increased as the bacterial population increased and as the amount of IF decreased, though the "bacterial B_{12} " could not be determined easily by the changes in the size of the bacterial population and IF.

(b) There was no significant difference in the "bacterial B_{12} " among the microorganisms examined (Fig. 3). The percent uptake of B_{12} seemed to reach the plateau of approximately 50 to 60% regardless of the increase in the bacterial population.

(c) The "bacterial B_{12} " was measured when incubated in distilled water and in the heart infusion broth which was prepared freshly. The "bacterial B_{12} " with fresh broth was greater than that incubated in distilled water (Table 3), while the turbidity of the former also increased from 480 to 720 during the 4 hour-incubation. On the contrary, there was no increase in the turbidity of the latter. In regard to the difference in the "bacterial B_{12} ", the increase in the bacterial population appeared not to be an important factor, because the ratios of B to A in Table 3 were practically constant.

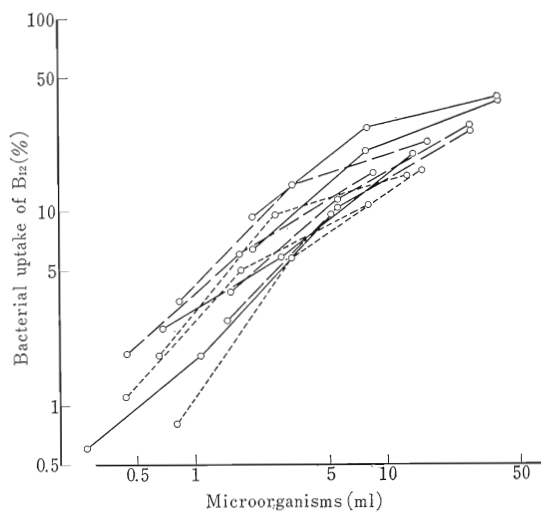


Fig. 3. Bacterial uptake of bound B₁₂. Mixtures of 1 m μ g of B₁₂ and 3 u of IF were incubated at 37°C for 60 minutes with *E. coli* (solid lines), *Klebsiella* (broken lines) and *Citrobacter* (dotted lines). Microorganisms were obtained in the early logarithmic phase of growth and in the later stage. The amount of microorganisms is expressed in ml of bacterial suspension with turbidity of 950.

(d) By changing the incubation temperature, the B₁₂ uptake by *Klebsiella* was observed (Table 4). The "bacterial B₁₂" increased as the incubation temperature was elevated and as the incubation time was prolonged. At 0°C the bacterial uptake was minimal and no increase in the uptake was observed with the prolongation of the incubation time. The ratios of the "bacterial B₁₂" at 37°C to the amount at 25°C were between 3.6 and 5.0.

Table 3. Bacterial uptake of bound B₁₂. *Klebsiella* (turbidity 480, 8 ml) and B₁₂-saturated IF (0.02 u) were incubated at 37°C in distilled water and heat infusion broth.

Incubation time (hours)	Bacterial uptake of B ₁₂ (%)		B/A
	In distilled water (A)	In heart infusion broth (B)	
1	13.0	17.8	1.37
2	25.1	33.3	1.33
3	32.1	46.5	1.45
4	39.9	56.0	1.40

Table 4. Bacterial uptake of bound B_{12} . *Klebsiella* (turbidity 270, 8 ml) was incubated with 0.02 u of B_{12} -saturated IF.

Incubation time (minutes)	Bacterial uptake of B_{12} (%)		
	Incubation temperature		
	0°C	25°C	37°C
30	4.5	4.7	16.9
60	4.6	6.4	32.2
120	4.2	9.9	44.2

On the other hand, these ratios ranged from 1.2 to 1.6 in the observation with 100 $m\mu g$ of free B_{12} as shown in Table 1.

(e) The effects of the IF concentrations on the bacterial uptake of B_{12} were observed by changing the order of the mixing of the three components (Fig. 4). The materials employed were *E. coli* (turbidity 230, 8 ml), 0.016 $m\mu g$ of B_{12} and 0.03 to 30.0 u of IF. The incubation was done at 37°C.

Experiment A: After incubation of *E. coli* with B_{12} for 1 hour, IF was added to the mixture and incubated again for another 1 hour. The "bacterial B_{12} " was practically constant regardless of the change in the amount of IF. It appeared that the B_{12} taken up by *E. coli* was not removed

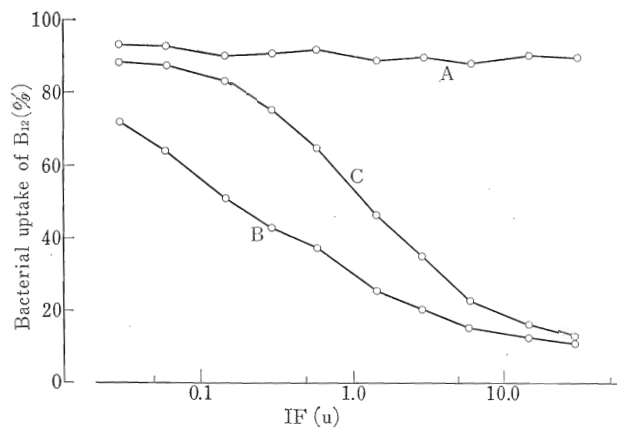


Fig. 4. Bacterial uptake of B_{12} . The bacterial uptake of B_{12} was observed by changing the order of mixing the three components as described in the text. The materials employed were *E. coli* (turbidity 230, 8 ml), 0.016 $m\mu g$ of B_{12} and 0.03 to 30.0 u of IF.

A: (*E. coli* + B_{12}) + IF, B: (B_{12} + IF) + *E. coli*, C: (*E. coli* + IF) + B_{12} .

by mixing a large amount of IF. Similar results were obtained with a larger amount of B₁₂ or with other microorganisms.

Experiment B: B₁₂ and IF were mixed for 10 minutes at room temperature and incubated for 2 hours with *E. coli*. The increase in the "bacterial B₁₂" was observed with the decrease in the amount of IF.

Experiment C: *E. coli* and IF were mixed for 5 minutes at room temperature and incubated with B₁₂ for 2 hours. The increase in the "bacterial B₁₂" was also observed with the decrease in the amount of IF. The "bacterial B₁₂" in Experiment C was always greater than that obtained in Experiment B. The microorganisms and the IF seemed to take up B₁₂ competitively in Experiment C, although there were few exceptions. One of the exceptions was that some calibrations might be necessary on these data, because the percent uptake of B₁₂ reached the plateau at approximately 10% and 90% instead of being flat at about 0% and 100%. Another exception was that the B₁₂-binding capacities of the microorganisms might be diminished to a 10th to 30th part, judging from the following findings: (i) The free B₁₂-binding capacity of these at time zero was suspected to be approximately 20 mμg, because the B₁₂-binding capacity at 1 hour was 30 mμg. (ii) In contrast, the B₁₂ was taken up by these microorganisms and IF equally, when the microorganisms and 1 to 2 u of IF were incubated in Experiment C.

(f) The bacterial uptake of bound B₁₂ was observed on two series of bound B₁₂ which had the same ratios of B₁₂ to IF. Bound B₁₂ (1:3) and bound B₁₂ (1:30) were prepared by mixing 1 volume of B₁₂ in 3 and 30 volumes of IF, respectively. The "bacterial B₁₂" increased along with the increase in B₁₂, and the percent uptake of B₁₂ decreased along with the increase in the bound B₁₂ (Fig. 5a). The patterns of the B₁₂ uptake with the different series of bound B₁₂ were not parallel.

The bacterial uptake of B₁₂ was expressed by the changes in the amount of free or total IF (Fig. 5b). Two-thirds to 100% of IF in bound B₁₂ (1:3) were estimated to be the amount of free IF in bound B₁₂ (1:3). Free IF in bound B₁₂ (1:30) was almost the same as the total IF in bound B₁₂ (1:30). The percent uptake of B₁₂ during incubation with bound (1:3) appeared to be identical in rate with bound B₁₂ (1:30). This finding suggested that the "bacterial B₁₂" was determined by the reciprocal changes in the size of the bacterial population and IF available.

Part 3. Nature of B₁₂ taken up by microorganisms

(a) The activity of IF was measured after the incubation of IF with microorganisms pre-incubated in the presence or absence of B₁₂. The B₁₂-

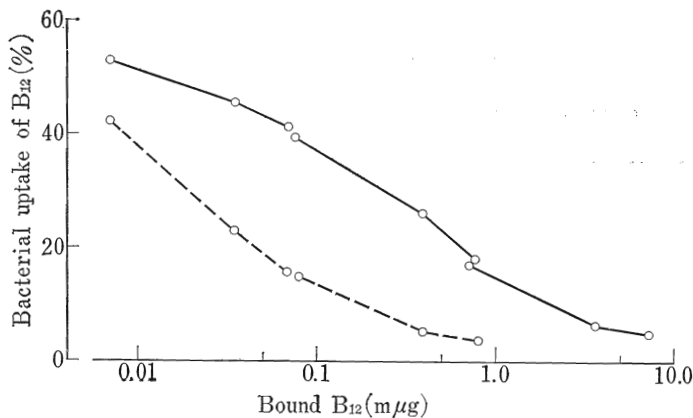


Fig. 5a. Bacterial uptake of bound B₁₂. *E. coli* (turbidity 240, 8 ml) was incubated with two series of bound B₁₂ at 37°C for 2 hours. Bound B₁₂ (1:3) and bound B₁₂ (1:30) were prepared by mixing 1 volume of B₁₂ in 3 and 30 volumes of IF, respectively.

— series of bound B₁₂ (1:3)
 - - series of bound B₁₂ (1:30)

binding capacity of IF occasionally decreased slightly, but this decrease was not in accordance with the increase in the bacterial population or the duration of incubation period. Therefore, it was not likely that the binding pattern of bound B₁₂ to microorganisms was B₁₂-IF-microorganism or IF-B₁₂-microorganism.

(b) B₁₂-saturated IF and *E. coli* were incubated at 37°C, 40°C and 45°C for 3 hours. The bacterial uptake of B₁₂ increased as the incubation temperature was elevated and as the incubation period was prolonged. Free IF activity in the supernatant increased along with the increase in the "bacterial B₁₂". It was also true when *Klebsiella* was incubated at 37°C and 45°C for 2 hours.

(c) Some of the above observations are shown in Table 5. The radioactivity of free IF in the supernatant increased in accordance with the increased "bacterial B₁₂". The B₁₂-binding capacity of free IF accounted for 96% to 98% of the total unsaturated B₁₂-binding capacity. The ratios of free IF (u) in the supernatant to the "bacterial B₁₂" (m μ g) ranged between 82% and 89%.

At time zero, the radioactivity was fairly small and the ratios obtained were not constant. The destruction of B₁₂-saturated IF was 6.6 to 9.5% during this observation and did not increase with the prolongation of incubation time. The B₁₂-binding capacity in the supernatant of the micro-

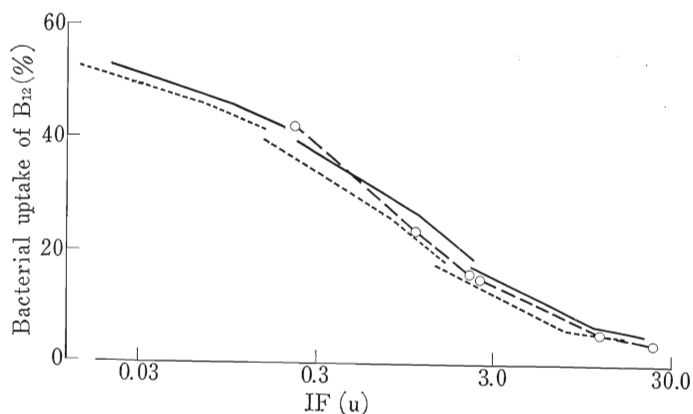


Fig. 5b. The results in Fig. 5a are calibrated by changing the abscissal scale from B₁₂ (m μ g) to IF (u).

- total IF in bound B₁₂ (1:3)
- - - - - 2/3 of IF in bound B₁₂ (1:3)
- · - - - total IF in bound B₁₂ (1:30)

Table 5. Correlation between the "bacterial B₁₂" and amount of free IF in the supernatant. B₁₂-saturated IF (0.125 u) was incubated with E. coli (turbidity 270, 5 ml) at 40°C for 3 hours. The radioactivity of 19,572 counts per 2 minutes was equivalent to 0.125 m μ g of B₁₂. The radioactivity of the background was approximately 100 counts and this value was already subtracted from the counts of the samples.

Incubation time (minutes)	B ₁₂ -binding capacity in supernatant of micro-organisms	Destruction of B ₁₂ -IF	Bacterial uptake of B ₁₂ (A)	B ₁₂ -binding capacity in supernatant			B-A (%)	(B/C)/A (%)
				Without IFA (B) ¹⁾	With IFA (C)	B-C ²⁾		
0	228	1,357	347	421	154	208	146	67
	238	1,233	274	483	134			
30		1,762	5,194	4,828	212	4,372	85	82
		1,936	5,514	4,322	192			
60	214	1,806	6,768	5,834	132	5,728	84	83
	141	1,689	6,927	5,917	162			
120		1,740	9,841	9,110	186	8,814	90	89
		1,795	9,943	8,868	163			
180	160	1,698	11,841	10,083	326	9,768	84	82
	144	1,709	11,929	9,898	219			

1) Radioactivity of total unsaturated B₁₂-binding capacity
 2) Radioactivity of B₁₂-binding capacity of free IF

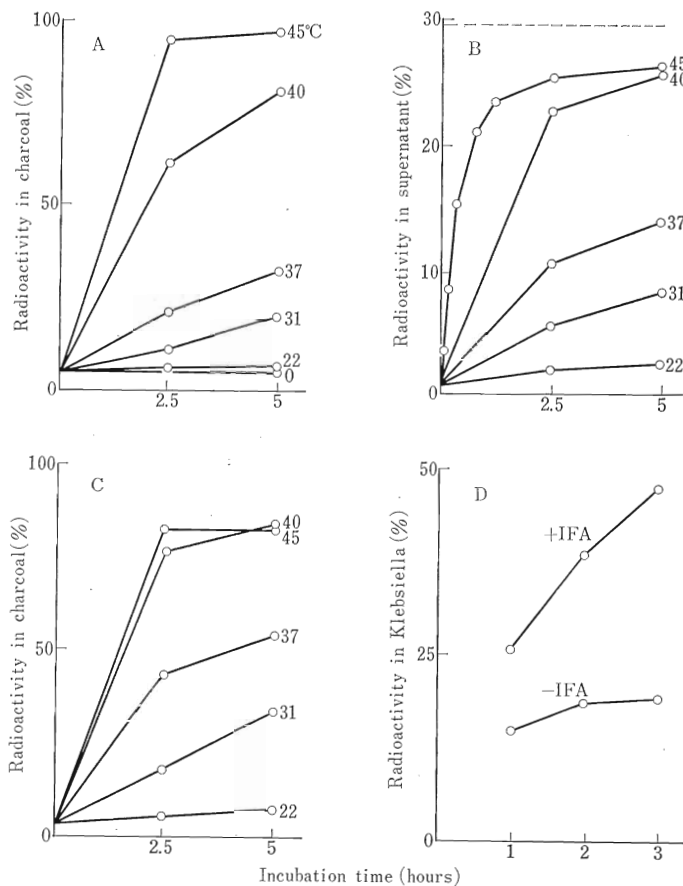


Fig. 6. Stability of B₁₂-IF complex. The rate of the dissociation of B₁₂ from IF was observed by different procedures as described in the text.

- Experiment A: labelled B₁₂-saturated IF + cold B₁₂
 Experiment B: cold B₁₂-saturated IF + labelled B₁₂
 Experiment C: labelled B₁₂-saturated IF + IFA
 Experiment D: labelled B₁₂-saturated IF + Klebsiella

organisms, after incubation in the absence of B₁₂-saturated IF, was also negligible.

Part 4. Stability of B₁₂-IF complex

In Experiment A, the radioactivity adsorbed on to the coated charcoal always increased as the incubation temperature was elevated and as the incubation time was prolonged (Fig. 6A). Similar findings were obtained in the supernatant of Experiment B. In these experiments, the destruction of IF appeared to be minimal, because the radioactivity in the supernatant

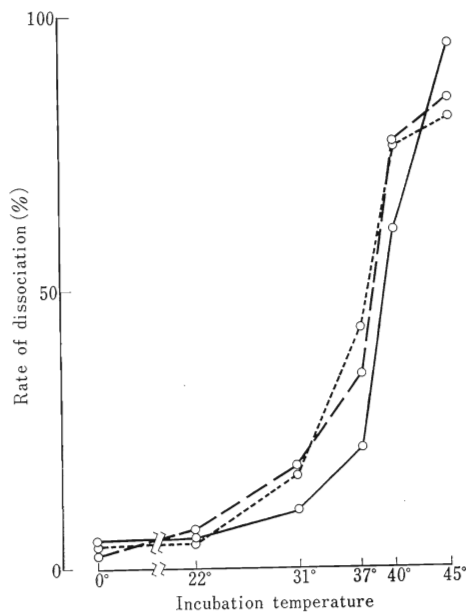


Fig. 7. Dissociation of B₁₂-IF complex. The rate of the dissociation by 150-minute incubation in Fig. 6 is summarized.

Experiment A —————
 Experiment B — — — — —
 Experiment C - - - - -

increased up to 25.7% by incubation at 45°C (Fig. 6B).

In Experiment C, the radioactivity also increased in the coated charcoal (Fig. 6C). It was suggested that the IF initially bound with labelled B₁₂ was liberated and lost its B₁₂-binding capacity after binding with IFA and that the labelled free B₁₂ derived from labelled B₁₂-saturated IF remained free of IF as the result of the decrease in the amount of IF.

Klebsiella (turbidity 630, 8 ml) and B₁₂-saturated IF (0.05 u) were incubated at 37°C for 3 hours with and without IFA (0.40 u). The "bacterial B₁₂" by incubation with IFA was greater than that without IFA (Fig. 6D). In this experiment, the "bacterial B₁₂" in the mixture without IFA reached the plateau after a 3 hour-incubation at which time the B₁₂-binding capacity of these microorganisms diminished to 70% of the initial value.

The rate of dissociation after 150-minute-incubation is summarized in Fig. 7. This rate sharply increased when the incubation temperature exceeded 37°C and the binding of B₁₂-IF was relatively stable below 31°C. This was also true when 1.4 u of B₁₂-saturated IF was employed.

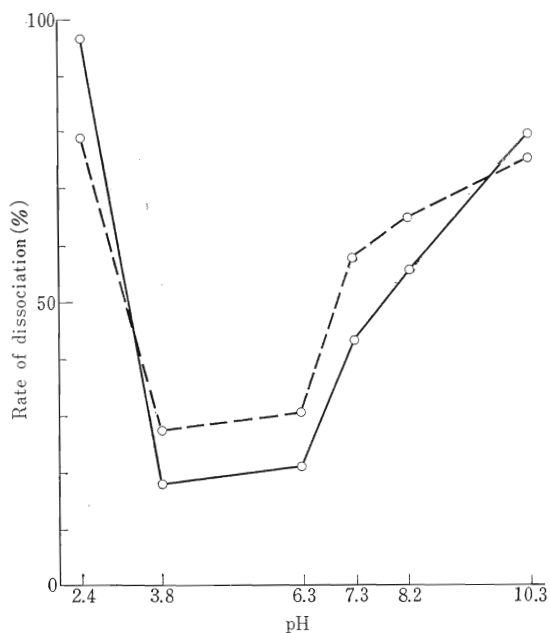


Fig. 8. Effects of pH on the dissociation of B_{12} -IF complex.

Solid lines: labelled B_{12} -saturated IF+cold B_{12}
 Broken lines: cold B_{12} -saturated IF+labelled B_{12}

Part 5. Effects of pH

The dissociation of B_{12} from IF was influenced by pH (Fig. 8). The binding of B_{12} -IF was most stable at around pH 3.8. In the range of pH higher than 6.3 the rate of dissociation increased. The rate of dissociation at pH 2.4 was greater than that at pH 3.8. The destruction of B_{12} -saturated IF (0.5 u) was also observed in a higher rate (21.3%) at this pH. The destruction was minimal (3.6 to 5.7%) at pH above 3.8.

The microorganisms (turbidity 400, 3 ml) were suspended in the same incubation media with various pH and incubated with free B_{12} (100 $m\mu\text{g}$) and B_{12} -saturated IF (1.4 u). The B_{12} -binding capacity of the microorganisms was lost almost completely at pH 2.4. There was no recognizable difference in the bacterial uptake of B_{12} between pH 3.8 and 10.3 (Fig. 9a), though small decreases in B_{12} -binding capacity of the microorganisms were observed at pH 10.3 in the present experiment and in the other experiments. On the other hand, the bacterial uptake of B_{12} initially bound to IF increased with the elevation of pH. This difference might be attributed to the increased dissociation of bound B_{12} along with the elevation of pH (Fig. 9b).

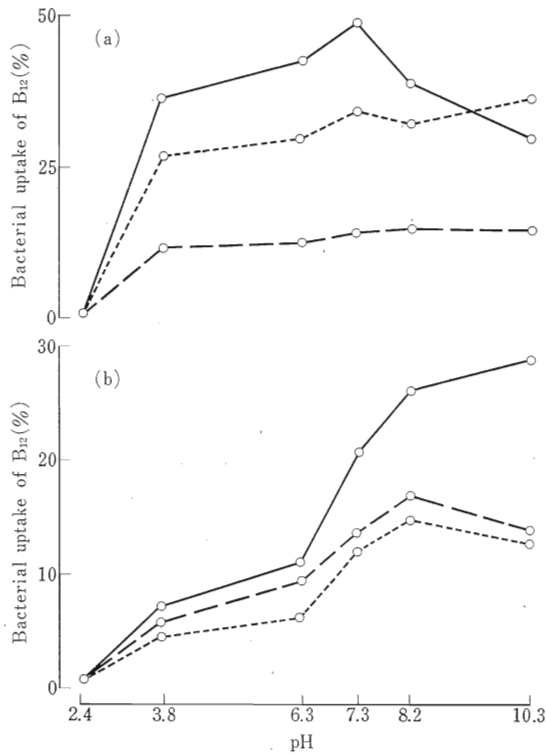


Fig. 9. Effects of pH on the bacterial uptake of B₁₂. Microorganisms (turbidity 400, 3 ml) were incubated at 37°C for 90 minutes with (a) 100 µg of B₁₂ and (b) 1.4 u of B₁₂-saturated IF.

E. coli ———, Klebsiella — — — —,
Citrobacter - - - - -

DISCUSSION

The intestinal flora took up free B₁₂ remarkably. Recently, it was reported that there is at least a two-stage process in the bacterial uptake of free B₁₂¹⁸). In the present observations, the bacterial uptake of B₁₂ occurred rapidly within 10 minutes and continued slowly up to 26 hours. It was also observed that the B₁₂ taken up by the microorganisms could not be removed, and the bacterial uptake of B₁₂ was partially inhibited by the pre-administration of another B₁₂. These results also suggested the two phases, slow adsorption after rapid adsorption, in the process of bacterial uptake of free B₁₂. A significant amount of B₁₂ was found in the microorganisms when bound B₁₂ and microorganisms were incubated. The

amount of B_{12} taken up by the microorganisms in the presence of IF was fairly smaller than that in the absence of IF.

As the next step, several attempts were made to examine the possibility that the microorganisms directly took up bound B_{12} (Part 3). No evidence to support this possibility was obtained, and it was questionable that the microorganisms disturbed IF when incubating in distilled water and saline. The reverse results were reported by Schjónsby and Tabaqchali¹⁹⁾ by incubation in the culture medium, although it is very difficult to measure the activity of IF precisely in the supernatant of the bacterial culture. This difference may come from the incubation medium.

In contrast, the increase in free IF activity in the supernatant was observed in accordance with the increase in the bacterial uptake of B_{12} , when B_{12} -saturated IF was incubated with the microorganisms (Table 5). It seemed that the free B_{12} could be taken up by the microorganisms after the liberation of B_{12} from IF, and that the IF was dissociated from B_{12} without a significant loss in its B_{12} -binding capacity.

There is an interesting report by Donaldson and Katz²⁶⁾ who demonstrated the exchanges between free and bound B_{12} . The stability of B_{12} -IF complex was influenced by the pH as well as by the temperature (Fig. 6 and 8). The ratios of the "bacterial B_{12} " at 37°C to the amount at 25°C when incubating with bound B_{12} were greater than those ratios when incubating with free B_{12} (Table 1 and 4). This phenomenon could be explained by the fact that the elevation of incubation temperature enhanced the dissociation of B_{12} from IF. The percent uptake of B_{12} reached the plateau regardless of the increase in the bacterial population (Fig. 3). The bacterial uptake of B_{12} was affected by pH (Fig. 9) as was observed by Gräsbeck²⁷⁾, though his results did not coincide with the authors' results. These observations suggested that the bacterial uptake of bound B_{12} is influenced by the dissociation of B_{12} from IF. On the other hand, it appeared that the B_{12} once taken up by the microorganisms could not be removed from the microorganisms and that the exchange between free B_{12} and the B_{12} taken up by the microorganisms could not occur.

The B_{12} -IF complex could not be taken up directly by the microorganisms, and the IF did not bind with the microorganisms nor with the B_{12} taken up by the microorganisms. These observations suggested that the microorganisms competed with the IF for the same point of B_{12} . This possibility was also supported by the results obtained in Fig. 4-C and 5.

In the present study, there was no suitable observation for the possibility that the microorganisms influenced the stability of the B_{12} -IF complex. However, this possibility could not be denied because the "bacterial

B₁₂" appeared occasionally to be greater than what was expected from the rate of dissociation of the B₁₂-IF complex.

The "bacterial B₁₂" in Experiment 5 was greater than that obtained in Experiment 3 in Table 2. However, when the ratio of free IF to the number of microorganisms was constant, the B₁₂-uptake per microorganisms was constant (Fig. 5). This discrepancy was not explained clearly in the present study.

It might be quite reasonable according to the above-mentioned hypothesis that malabsorption of B₁₂ was occasionally improved by the administration of IF in the patients^{1,4,8)} and in the experimental animals²³⁾. In these cases, by the supplementation of IF, it appeared that the concentration of IF became sufficient to compete with the microorganisms. On the other hand, there were many cases of B₁₂ malabsorption which appeared not to be improved by the supplementation of IF but improved after the administration of antibiotics. In such cases, it seemed that there was so large a number of microorganisms in the blind loop that the supplementation of IF appeared to be of no effect. In the present investigation, a large quantity of additional IF was necessary to overwhelm the activity of the microorganisms to take up B₁₂ (Fig. 4 and 5).

There were approximately 10⁴ of microorganisms per ml in the contents of the small intestine in the normal subjects²⁸⁻³¹⁾, although it has been considered that the lumen was practically sterile. On the other hand, no suitable explanation has been given to the fact that there were significant amounts of B₁₂ (from 20 to 40%) which were not absorbed, when small doses of B₁₂ were administered to the normal subjects. If the unabsorbed B₁₂ is attributed to the microorganisms as the resident in the lumen, B₁₂ malabsorption in the blind loop syndrome will be explainable by itself.

All microorganisms, which possessed the B₁₂-binding capacity, showed almost the same properties as shown in the present investigation. Therefore, B₁₂ malabsorption can be induced by the microorganisms which possess a B₁₂-binding capacity.

REFERENCES

- 1) Badenoch, J., Bedford, P. D., and Evans, J. R.: Massive diverticulosis of the small intestine with steatorrhoea and megaloblastic anaemia. *Quart. J. Med.*, 24: 321-330, 1955.
- 2) Halsted, J. A., Lewis, P. M., and Gasster, M.: Absorption of radioactive vitamin B₁₂ in the syndrome of megaloblastic anemia associated with intestinal stricture or anastomosis. *Amer. J. Med.*, 20: 42-52, 1956.
- 3) Gellman, D. D.: Diverticulosis of the small intestine with steatorrhea and megaloblastic anemia. *Am. J. Surg.*, 1956.

- blastic anemia. *Lancet*, 2: 873-874, 1956.
- 4) Scudamore, H. H., Hagedorn, A. B., Wollaeger, E. E., and Owen, C. A.: Diverticulosis of the small intestine and macrocytic anemia with report of two cases and studies on absorption of radioactive vitamin B₁₂. *Gastroenterology*, 34: 66-82, 1958.
 - 5) Doig, A., and Girdwood, R. H.: The absorption of folic acid and labelled cyanocobalamin in intestinal malabsorption. *Quart. J. Med.*, 29: 334-374, 1960.
 - 6) Polachek, A. A., Pijanowski, W. J., and Miller, J. M.: Diverticulosis of the jejunum with macrocytic anemia and steatorrhea. *Ann. Intern. Med.*, 54: 636-645, 1961.
 - 7) Schiffer, L. M., Faloan, W. W., Chodos, R. B., and Lozner, E. L.: Malabsorption syndrome associated with intestinal diverticulosis. *Gastroenterology*, 42: 63-68, 1962.
 - 8) Cook, W. T., Cox, E. V., Fone, D. J., Meynell, M. J., and Gaddie, R.: The clinical and metabolic significance of jejunal diverticula. *Gut*, 4: 115-131, 1963.
 - 9) Wirts, C. W., and Goldstein, F.: Studies of the mechanism of postgastrectomy steatorrhea. *Ann. Intern. Med.*, 58: 25-36, 1963.
 - 10) Paulk, E. A., and Farrar, W. E.: Diverticulosis of the small intestine and megaloblastic anemia. *Amer. J. Med.*, 37: 473-480, 1964.
 - 11) Polter, D. E., Boyle, J. D., Miller, L. G., and Finegold, S. M.: Anaerobic bacteria as cause of the blind loop syndrome. *Gastroenterology*, 54: 1148-1154, 1968.
 - 12) Ternberg, J. L., and Eakin, R. E.: Erythrin and apoerythrin and their relation to the antipernicious anemia principle. *J. Amer. Chem. Soc.*, 71: 3858, 1949.
 - 13) Burkholder, P. R.: Microbiological studies on materials which potentiate oral vitamin B₁₂ therapy in Addisonian anemia. *Arch. Biochem.*, 39: 322-332, 1952.
 - 14) Booth, C. C., and Heath, J.: The effect of *E. coli* on the absorption of vitamin B₁₂. *Gut*, 3: 70-73, 1962.
 - 15) Donaldson, R. M., Corrigan, H., and Natosios, G.: Malabsorption of Co⁶⁰-labelled cyanocobalamin in rats with intestinal diverticula. II. Studies on contents of the diverticula. *Gastroenterology*, 43: 282-290, 1962.
 - 16) Sherwood, W. C., Goldstein, F., Haurani, F. I., and Wirts, C. W.: Studies of the small-intestinal bacterial flora and of intestinal absorption in pernicious anemia. *Amer. J. Dig. Dis.*, 9: 416-425, 1964.
 - 17) Dellipiani, A. W., Samson, R. R., and Girdwood, R. H.: The uptake of vitamin B₁₂ by *E. coli*: Possible significance in relation to the blind loop syndrome. *Amer. J. Dig. Dis.*, 13: 718-726, 1968.
 - 18) Giannella, R. A., Broitman, S. A., and Zamchek, N.: Vitamin B₁₂ uptake by intestinal microorganisms: Mechanism and relevance to syndromes of intestinal bacterial overgrowth. *J. Clin. Invest.*, 50: 1100-1107, 1971.
 - 19) Schjónsby, H., and Tabaqchali, S.: Effect of small intestinal bacteria on intrinsic factor and the vitamin B₁₂-intrinsic factor complex. *Scand. J. Gastroenterol.*, 6: 707-713, 1971.
 - 20) Giannella, R. A., Broitman, S. A., and Zamchek, N.: Competition between bacteria and intrinsic factor for vitamin B₁₂: Implications for vitamin B₁₂ malabsorption in intestinal bacterial overgrowth. *Gastroenterology*, 62: 255-260, 1972.
 - 21) Watson, G. M., and Witts, L. J.: Intestinal macrocytic anaemia. *Brit. Med. J.*, i: 13-17, 1952.
 - 22) Strauss, E. W., Donaldson R. M., and Gardner, F. H.: A relationship between intestinal bacteria and the absorption of vitamin B₁₂ in rats with diverticula of the small bowel. *Lancet*, 2: 736-738, 1961.
 - 23) Donaldson, R. M.: Malabsorption of Co⁶⁰-labeled cyanocobalamin in rats with intestinal diverticula. I. Evaluation of possible mechanisms. *Gastroenterology*, 43: 271-281, 1962.

- 24) Goldstein, F., Cozzolino, H. J., and Wirts, C. W.: Diarrhea and steatorrhea due to a large solitary duodenal diverticulum: Report of a case. *Amer. J. Dig. Dis.*, 8: 937-943, 1963.
- 25) Gottlieb, C., Lau, K.-S., Wasserman, L. R., and Herbert, V.: Rapid charcoal assay for intrinsic factor (IF), gastric juice unsaturated B₁₂ binding capacity, antibody to IF, and serum unsaturated B₁₂ binding capacity. *Blood*, 25: 875-884, 1965.
- 26) Donaldson, R. M., and Katz, J. H.: Exchange between free and gastric juice-bound cyanocobalamin. *J. Clin. Invest.*, 42: 534-545, 1963.
- 27) Gräsbeck, R.: The vitamin B₁₂-binding principle of human gastric juice: Influence of pH on the bacterial adsorption of free and bound B₁₂. *Scand. J. Clin. Lab. Invest.*, 9: 50-53, 1957.
- 28) Kalsner, M. H., Cohen, R., Arteaga, I., Yawn, E., Mayoral, L., Hoffert, W. R., and Frazier D.: Normal viral and bacterial flora of the human small and large intestine. *New Engl. J. Med.*, 274: 500-505, 558-563, 1966.
- 29) Gorbach, S. L., Plaut, A. G., Nahas, L., Weinstein, L., Spanknebel, G., and Levitan, R.: Studies of intestinal microflora. II. Microorganisms of the small intestine and their relations to oral and fecal flora. *Gastroenterology*, 53: 856-867, 1967.
- 30) Plaut, A. G., Gorbach, S. L., Nahas, L., Weinstein, L., Spanknebel, G., and Levitan, R.: Studies of intestinal microflora. III. The microbial flora of human small intestinal mucosa and fluids. *Gastroenterology*, 53: 868-873, 1967.
- 31) Drasar, B. S., Shiner, M., and McLeod, G. M.: Studies on the intestinal flora. I. The bacterial flora of the gastrointestinal tract in healthy and achlorhydric persons. *Gastroenterology*, 56: 71-79, 1969.